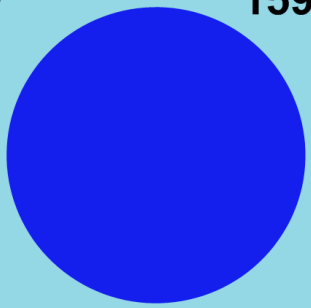





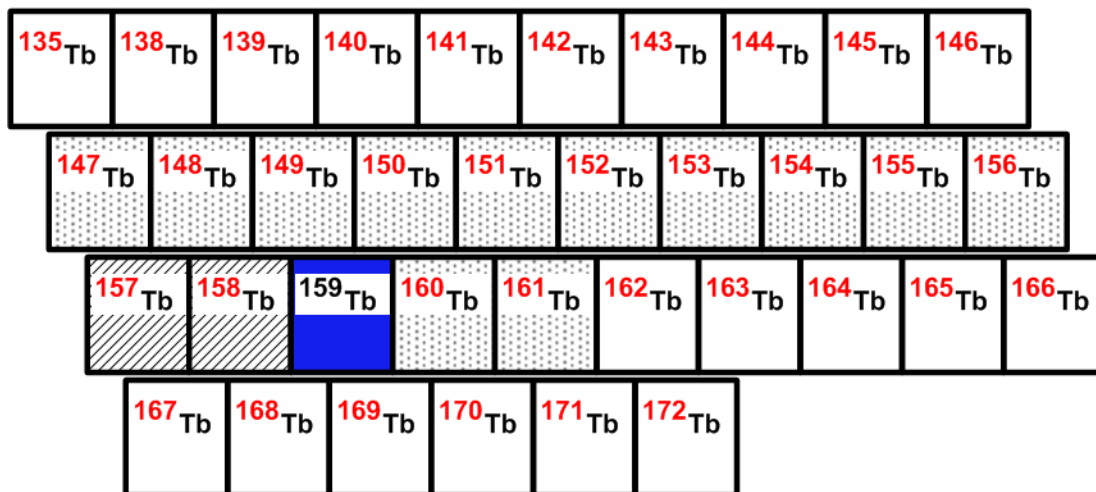
4.65 terbium

terbium Tb 65  158.925 354(8)
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Stable isotope	Relative atomic mass	Mole fraction
^{159}Tb	158.925 35	1

Half-life of radioactive isotope

Less than 1 hour	
Between 1 hour and 1 year	
Greater than 1 year	



4.65.1 Terbium isotopes in medicine

^{149}Tb (with a **half-life** of 4.1 hours) is being used in targeted **radiotherapy** using **alpha particles** for labeling **radioimmunoconjugates** in cancer treatments [455, 456]. ^{161}Tb (with a half-life of 6.9 days) attached to a bioconjugate (two **covalently** linked molecules, one or more of which is a biomolecule), is being used in cancer therapy as a targeted radiation treatment of cancer cells [456, 457]. ^{161}Tb is being used for imaging as it allows for on-line monitoring of its distribution using **gamma cameras** [457]. ^{149}Tb is produced by the reaction $^{142}\text{Nd}(^{12}\text{C}, 5n)^{149}\text{Dy} \rightarrow ^{149}\text{Tb} + \beta^+$

and by $^{141}\text{Pr}(^{12}\text{C},4n)^{149}\text{Tb}$, and beam geometry is important for satisfactory yield of ^{149}Tb (Figure 4.65.1) [458].

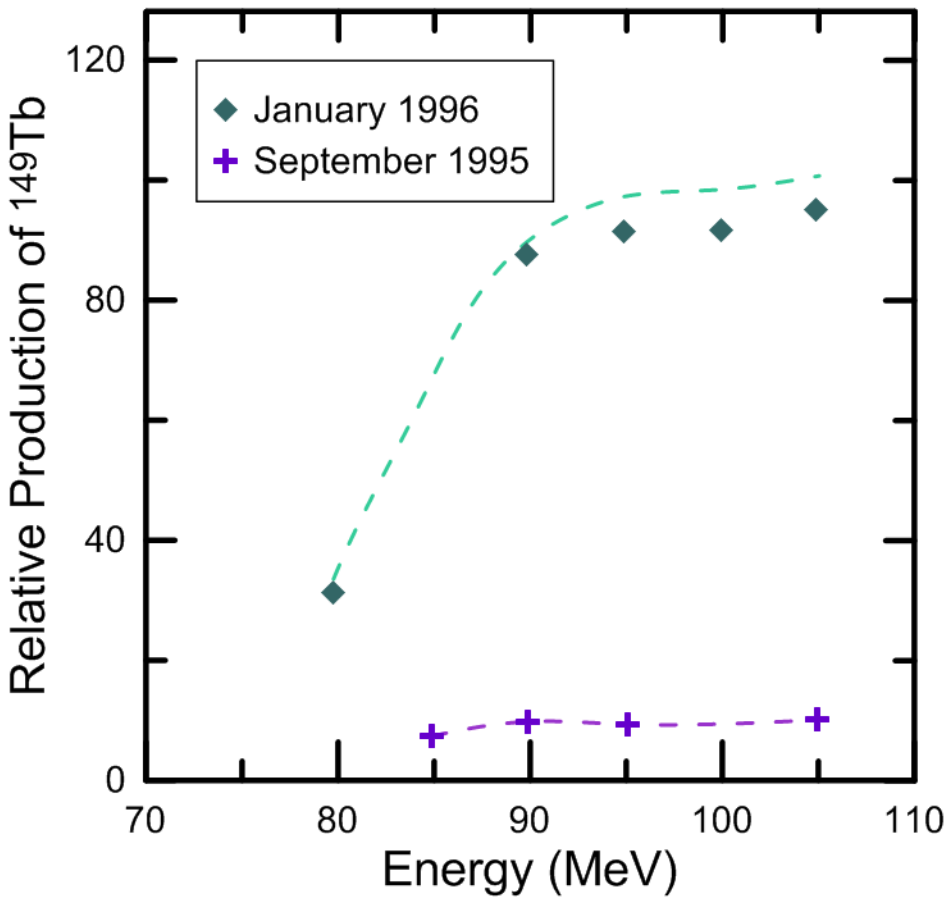


Fig. 4.65.1: Relative production of ^{149}Tb from the reaction $^{142}\text{Nd}(^{12}\text{C},5n)^{149}\text{Dy} \rightarrow ^{149}\text{Tb} + \beta^+$ for two different beam geometries. A ten-fold increase in production is achieved by optimal beam geometry (January 1996) (modified from [458]).